

# General Chemistry I

## CHM 1045

### Course Description:

Prerequisite: MAC 1105 with a grade of "C-" or better or placement beyond MAC 1105. This course includes topics such as chemical symbols, formulas, and equations; states of matter; reactivity in aqueous solution; electronic structure, bonding, and molecular geometry. Students taking CHM 1045 after taking CHM 1020 and/or CHM 1032 may register for reduced credit, as indicated in the department's policy on reduced credit.

**Course Learning Objectives** (including liberal studies learning objectives assigned to the course):

Upon completion of this course, students will be able to:  
generally

- Use chemistry concepts and terms to describe everyday phenomena.
- Use scientific thinking to solve real-world problems.
- Appreciate the types of problems chemistry can be used to solve.
- Engage in inquiry (especially by asking why and how questions).
- Evaluate the reasonableness of your own and other people's answers to questions.
- Make reasonable estimates and recognize trends.
- Use images, diagrams, and structural models to represent chemical phenomena.
- Think critically and cogently about causal relationships with scientific reasoning.
- Assess previous experimentation and published scientific results.
- Critically examine and evaluate scientific observation, hypothesis or model construction.
- Articulate a variety of issues created by the complex interactions among science, technology, and society.
- Use scientific perspectives to evaluate contemporary problems facing society.
- Translate real-world observations into chemical terms, including by translating an observation to symbolic representation (e.g., heat to temperature) and describing chemical relationships using symbolic representations, and making decisions (i.e., conclusions) based on those relationships.

and more specifically

- Differentiate between accuracy and precision in measurements.
- Report measurements and calculations to the proper number of significant figures.
- Solve problems using the Unit-Label/Dimensional Analysis method.
- Discuss the structure of the atom and organization of the periodic table.
- Name and write formulas for ionic and molecular compounds.
- Solve problems using mole relationships.
- Balance chemical equations and predict products of chemical reactions.
- Use balanced chemical equations to solve stoichiometry problems.
- Write net ionic equations for double replacement reactions.

- Identify and balance redox reactions using the half - reaction method.
- Solve problems using solution concentration (molarity).
- Solve problems using the gas laws.
- Solve problems using the First Law of Thermodynamics.
- Relate the spectrum of an element to the structure of the atom.
- Write the electron configuration for an element and relate to the structure of the atom.
- Name and identify the four quantum numbers for an element.
- Predict properties of elements based on the trends of the periodic table.
- Differentiate between ionic and covalent bonding.
- Draw Lewis structures and predict molecular geometry of a molecule.
- Predict and explain the polarity of a molecule.
- Explain the geometry of a molecule using one of the bonding theories.

### **Grading Assignments:**

Homework; quizzes; worksheets; participation assignments; chemistry concept surveys; examinations; final examination; liberal studies assignment

### **LIBERAL STUDIES STATEMENTS**

This course has been approved to meet FSU's Liberal Studies Natural Sciences requirement and helps you become an effective interpreter of scientific results and a critical analyst of claims about the natural world.

By the end of this course, students will:

1. Pose questions or hypotheses based on scientific principles.
2. Use appropriate scientific methods and evidence to evaluate claims or theoretical arguments about the natural world.
3. Analyze and interpret research results using appropriate methods.

### **UNIVERSITY ATTENDANCE POLICY**

Excused absences include documented illness, deaths in the family and other documented crises, call to active military duty or jury duty, religious holy days, and official University activities. These absences will be accommodated in a way that does not arbitrarily penalize students who have a valid excuse. Consideration will also be given to students whose dependent children experience serious illness.

### **ACADEMIC HONOR POLICY**

The Florida State University Academic Honor Policy outlines the University's expectations for the integrity of students' academic work, the procedures for resolving alleged violations of those

expectations, and the rights and responsibilities of students and faculty members throughout the process. Students are responsible for reading the Academic Honor Policy and for living up to their pledge to "...be honest and truthful and... [to] strive for personal and institutional integrity at Florida State University." (Florida State University Academic Honor Policy, found at <http://fda.fsu.edu/Academics/Academic-Honor-Policy>)

## **AMERICANS WITH DISABILITIES ACT**

Florida State University (FSU) values diversity and inclusion; we are committed to a climate of mutual respect and full participation. Our goal is to create learning environments that are usable, equitable, inclusive, and welcoming. FSU is committed to providing reasonable accommodations for all persons with disabilities in a manner that is consistent with academic standards of the course while empowering the student to meet integral requirements of the course.

To receive academic accommodations, a student:

(1) must register with and provide documentation to the Office of Accessibility Services (OAS); (2) must provide a letter from OAS to the instructor indicating the need for accommodation and what type; and, (3) should communicate with the instructor, as needed, to discuss recommended accommodations. A request for a meeting may be initiated by the student or the instructor.

Please note that instructors are not allowed to provide classroom accommodations to a student until appropriate verification from the Office of Accessibility Services has been provided. This syllabus and other class materials are available in alternative format upon request. For more information about services available to FSU students with disabilities, contact the

Office of Accessibility Services  
874 Traditions Way  
108 Student Services Building  
Florida State University, Tallahassee, FL 32306-4167  
(850) 644-9566 (voice) (850) 644-8504 (TDD)  
[oas@fsu.edu](mailto:oas@fsu.edu)  
<https://dsst.fsu.edu/oas>

## **ACADEMIC SUCCESS**

Your academic success is a top priority for Florida State University. University resources to help you succeed include tutoring centers, computer labs, counseling and health services, and services for designated groups, such as veterans and students with disabilities. The following information is not exhaustive, so please check with your advisor or the Dean of Students office to learn more.

## **CONFIDENTIAL CAMPUS RESOURCES**

Various centers and programs are available to assist students with navigating stressors that might impact academic success. These include the following:

Victim Advocate Program

University Center A, Room 4100  
(850) 644-7161 Available 24/7/365  
Office Hours: M-F 8-5  
<https://dsst.fsu.edu/vap>

Counseling & Psychological Services  
Askew Student Life Center, 2nd Floor  
942 Learning Way  
(850) 644-8255  
<https://counseling.fsu.edu/>

University Health Services  
Health and Wellness Center  
(850) 644-6230  
<https://uhs.fsu.edu>